

WEBINAR

# Washout Considerations for ICP Analyses

THURSDAY, MARCH 11  
9:00–10:00AM EST

## SPEAKERS:

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Director,  
Production



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# Key Topics

- Causes for washout issues
- Introduction system components
- Elements of concern
- Common rinse solutions / strategies
- Maintenance recommendations



# Audience Participation

- Today we are going to poll the audience throughout the presentation to better understand our collective challenges in the lab.
- Most of these questions will be multiple choice, and we will get to see the results after each question.
- Don't fall asleep!



# Question 1

- What type of instrument do you use in your laboratory?

- A. ICP-OES
- B. ICP-MS
- C. Both ICP-OES & ICP-MS



# Causes for washout issues

- Solution matrix ( $\text{HNO}_3/\text{HCl}/\text{HF}/\text{NH}_4\text{OH}$ )
  - Could be from commercially available standards
  - Could be from specific sample preparation protocols
- Specific elemental affinity toward different types of plastics (Pump tubing, HF-resistant intro system)
- Spray chamber design (Single-pass, double-pass)
- Rinse protocols using the wrong acids/bases
- Allowing for the appropriate amount of time to rinse



# Solution Matrix

- Most elements are stable using only  $\text{HNO}_3$
- Some elements require HF for stability
  - HF stability often revolves around plastic surfaces
  - Some elements will precipitate in the presence of HF
- Some elements require HCl for stability
  - Only a few elements have issues in the presence of HCl
- Bromide/Iodide require basic matrices for stability
  - Some other elements are stable in base without  $\text{HNO}_3$ /HF



## Question 2

- Does your company forbid the use of HF in the laboratory?

A. Yes – HF is forbidden

B. No – I can use HF



# Solution Matrix - HF

H			Not Checked by ICP										HF Elements					He
Li	Be		HF "thieves"										B	C	N	O	F	Ne
Na	Mg												Al	Si	P	S	Cl	Ar
K	Ca		Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr		Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	*	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	**	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og

*	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
**	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No





# Solution Matrix - HCl

H	Not Checked by ICP																HCl Elements						He						
Li	Be	Can work w/o HF																Avoid HCl						B	C	N	O	F	Ne
Na	Mg																	Al	Si	P	S	Cl	Ar						
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr												
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe												
Cs	Ba	*	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn											
Fr	Ra	**	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og											

*	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
**	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No



# Solution Matrix – TEA/NH<sub>4</sub>OH/H<sub>2</sub>O

H		Not Checked by ICP										Basic Elements					He	
Li	Be	Can work w/o HNO <sub>3</sub> /HF, but must be basic										B	C	N	O	F	Ne	
Na	Mg											Al	Si	P	S	Cl	Ar	
K	Ca		Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr		Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	*	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	**	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og

*	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
**	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No



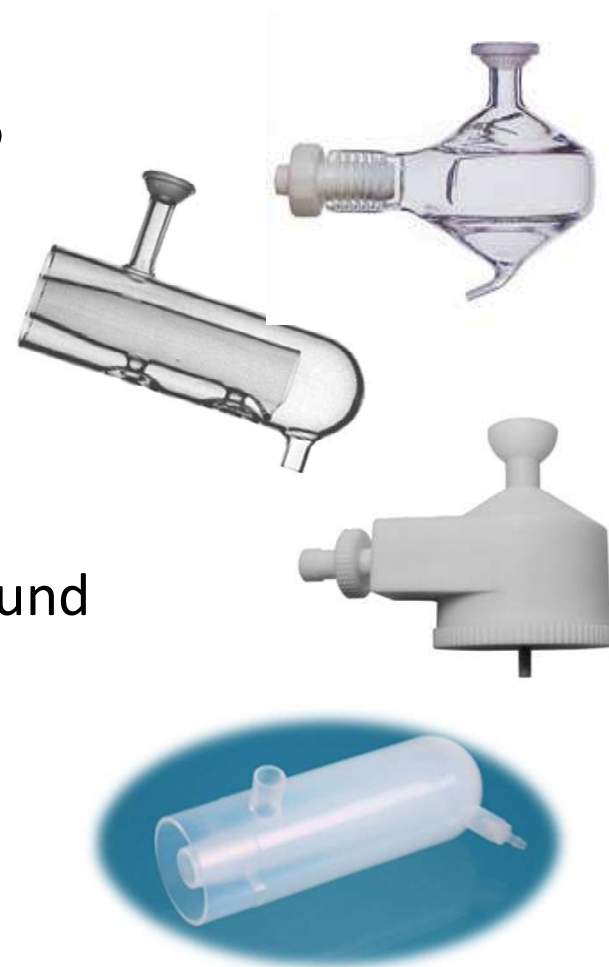
# Key Topics

- Causes for washout issues
- **Introduction system components**
- Elements of concern
- Common rinse solutions / strategies
- Maintenance recommendations



# Spray Chambers

- Classical borosilicate glass
  - If HF is used, problems testing B and Si
  - B and Si will leach out of the spray chamber
- HF Resistant Systems (PTFE/PFA)
  - If high levels are run, B, Si, and Hg can stick around
  - No leaching of B and Si from the material
  - Coating is essential to help with performance
- Double pass spray chambers increase washout time significantly



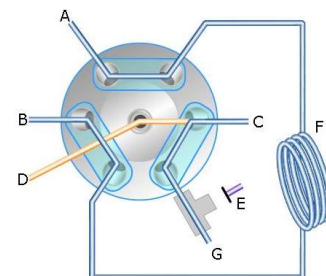
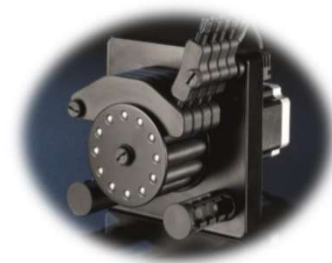
## Question 3

- What type of spray chamber do you use on your primary ICP?
- A. Borosilicate Glass Cyclonic
  - B. Borosilicate Glass Scott Style
  - C. HF-Resistant Teflon Cyclonic
  - D. HF-Resistant Teflon Scott Style



# Sample Delivery

- **Peristaltic Pump**
  - Sample is introduced from autosampler probe through peristaltic pump tubing before it reaches the nebulizer
  - Many elements stick to the PVC peristaltic pump tubing
- **Syringe Drive / Switching Valve Systems**
  - Sample is loaded into a sample loop without passing through peristaltic pump tubing
  - PVC tubing is eliminated, but a switching valve is added to the equation



## Question 4

- What type of sample delivery system do you use on your primary ICP?
- A. Standard Peristaltic Pump
  - B. Peristaltic Pump with Switching Valve
  - C. Syringe Drive with Switching Valve



# Peri-Pump vs. Syringe Drive on MS

	Blank1	Blank2	Blank3	Sample	+4ppb Spike	Blank4	Blank5	Blank6	
95Mo	6	6	12	1,229	28,443	<b>297</b>	<b>116</b>	<b>77</b>	Peri Pump
95Mo	17	11	13	756	28,443	<b>56</b>	<b>15</b>	<b>15</b>	Syringe Drive
121Sb	17	9	23	437	70,605	<b>1,547</b>	<b>731</b>	<b>443</b>	Peri Pump
121Sb	9	2	4	193	70,605	<b>7</b>	<b>2</b>	<b>11</b>	Syringe Drive
178Hf	36	18	26	5	134,673	<b>134</b>	<b>87</b>	<b>64</b>	Peri Pump
178Hf	0	0	0	4	134,673	<b>11</b>	<b>2</b>	<b>4</b>	Syringe Drive
181Ta	34	55	43	27	461,654	<b>1,467</b>	<b>801</b>	<b>514</b>	Peri Pump
181Ta	20	13	11	7	461,654	<b>87</b>	<b>20</b>	<b>29</b>	Syringe Drive
184W	85	58	58	394	127,417	<b>2,876</b>	<b>1,097</b>	<b>669</b>	Peri Pump
184W	35	26	44	150	127,417	<b>420</b>	<b>226</b>	<b>141</b>	Syringe Drive
209Bi	78	75	50	131	271,275	<b>7,664</b>	<b>2,590</b>	<b>1,323</b>	Peri Pump
209Bi	18	18	31	51	271,275	<b>38</b>	<b>22</b>	<b>24</b>	Syringe Drive
232Th	5	35	12	1,467	349,163	<b>478</b>	<b>173</b>	<b>163</b>	Peri Pump
232Th	15	7	6	15	349,163	<b>26</b>	<b>30</b>	<b>15</b>	Syringe Drive

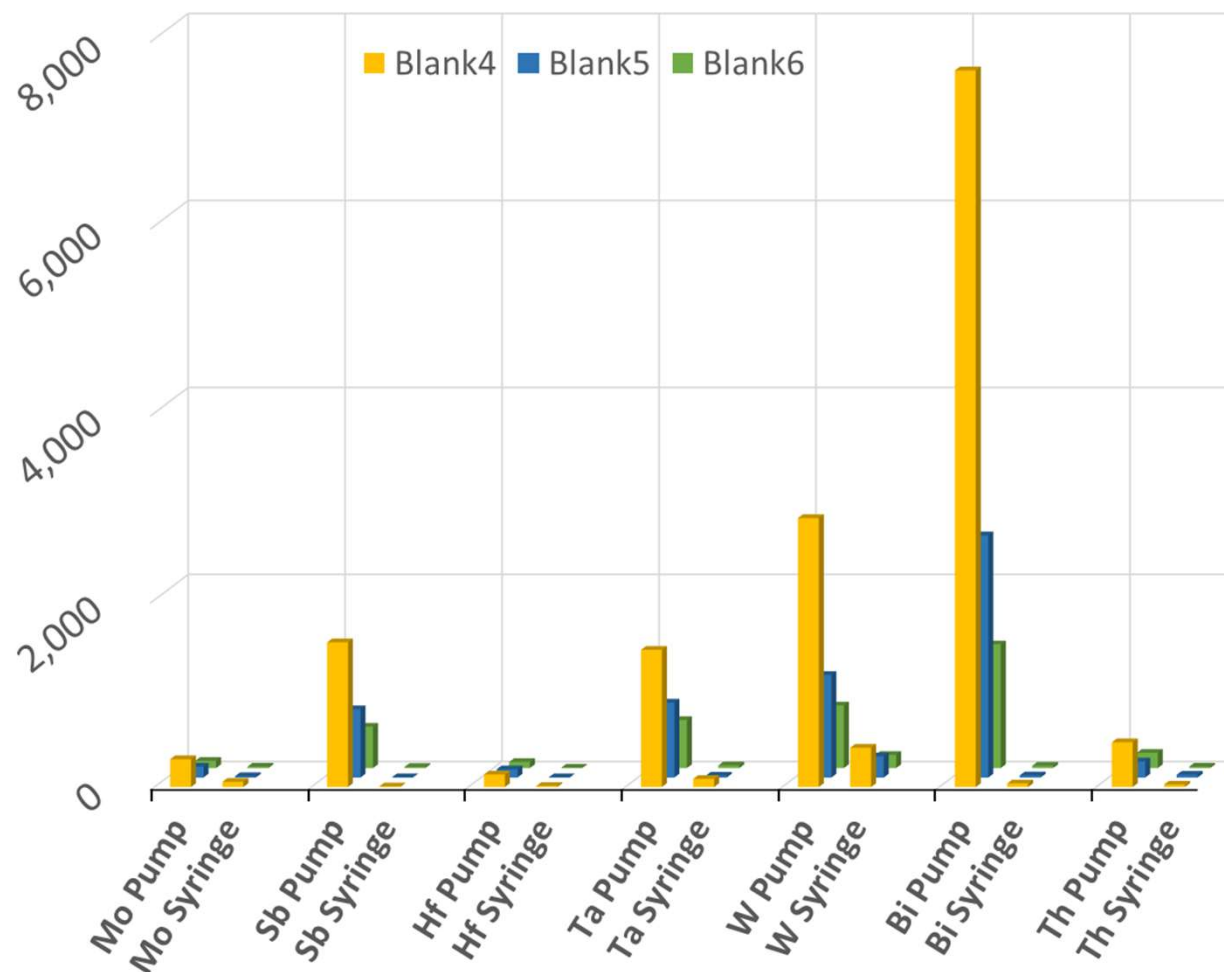
The sample run in this example is 100µg/g Mn in 1% v/v HNO<sub>3</sub>.



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- Washout of select “sticky” elements after a 4ppb spike containing over 60 elements.
- Bi is by far the worst.
- PVC tubing is used for peri-pump introduction systems.
- Faster washout of elements using syringe drive systems allows us to run more samples free of “memory” interferences.
- This results in less maintenance when running high TDS samples.



# Key Topics

- Causes for washout issues
- Introduction system components
- **Elements of concern**
- Common rinse solutions / strategies
- Maintenance recommendations



# Problem Elements

H	Not Checked by ICP																Elements with washout issues					He	
Li	Be																	B	C	N	O	F	Ne
Na	Mg																	Al	Si	P	S	Cl	Ar
K	Ca		Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr					
Rb	Sr		Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe					
Cs	Ba	*	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn					
Fr	Ra	**	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og					

*	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
**	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No



# Why?

H	Not sure, PVC tubing?																HNO <sub>3</sub> makes it sticky																He						
Li	Be	Lack of HF makes it sticky																HCl makes it precipitate																B	C	N	O	F	Ne
Na	Mg																	Al	Si	P	S	Cl	Ar																
K	Ca		Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr																					
Rb	Sr		Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe																					
Cs	Ba	*	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn																					
Fr	Ra	**	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og																					

*	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
**	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No



# Are you running elements you don't need?

- Custom solutions can help you rid your analyses of unnecessary problematic elements
- Concentration ranges can be high (for user dilution)
- Or they can be ready made to save time at the bench
- IV can ensure that your solutions will be stable and suitable for your methods
- We can even make recommendations for elemental/matrix compatibility



# Key Topics

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# Common Rinse Solutions

- $\text{HNO}_3$ 
  - 5-10% on an OES
  - 1-2% on a MS
- $\text{HCl}$ 
  - 5-10% on an OES
  - 1-2% on a MS
- RBS-25
  - 2.5% on an OES
  - Not Recommended on MS due to high Sodium
- $\text{H}_2\text{O}$ 
  - Can be effective enough for Na, K, Ca, etc.



# Rinse Solutions with HF

- $\text{HNO}_3$

- 5-10% on an OES
- 1-2% on a MS

- HF

- 0.1-2% on an OES
- 0.05-0.5% on a MS

- If using borosilicate glass nebulizer and spray chamber



- Limit HF to a max of 0.2%
- B and Si results will be unreliable

- If using an HF resistant nebulizer and spray chamber



- Can go up to 2-3%
- >3% HF will degrade the coating





# Specialty Rinse Solutions

- $\text{NH}_4\text{OH}$

- 1-5% for OES or MS
- Use for B, Br, I, Hg

- $\text{HCl}$  / Thiourea

- 1-10%  $\text{HCl}$
- 0.5% Thiourea
- Use for Hg, Au, Os

- $\text{HCl}$  / Hydroxylamine· $\text{HCl}$

- 10%  $\text{HCl}$
- 0.5%  $\text{NH}_2\text{OH}\cdot\text{HCl}$
- Use for Os

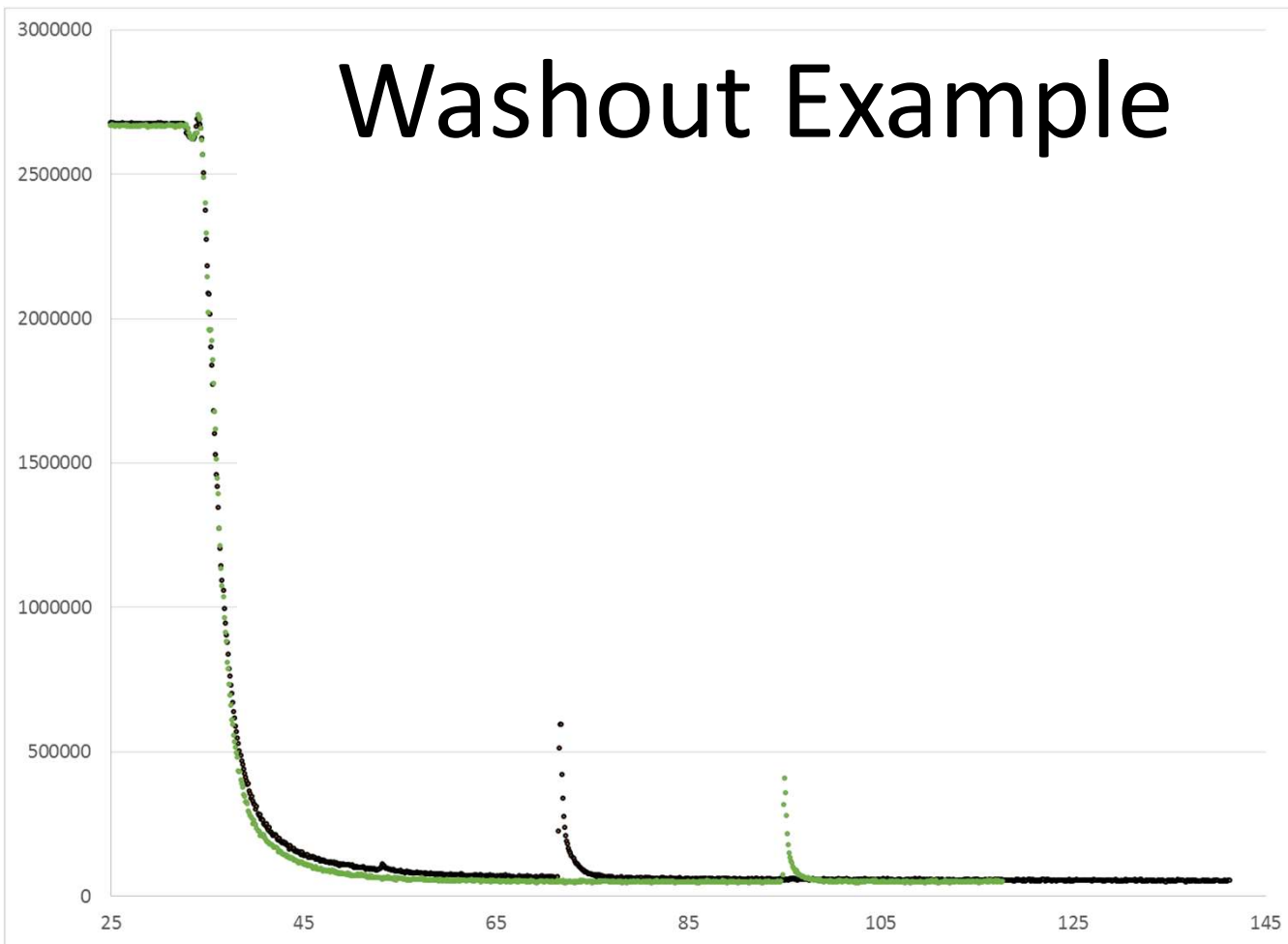


# Utilize real-time output monitoring

- Some instrument software packages allow you to monitor a line in real time and record the signal vs time
- This can help with method development activities in determining appropriate rinse times between samples
- It can also be helpful in assessing the effectiveness of different rinse solutions



# Washout Example



100ppm Boron in  
0.1% v/v HNO<sub>3</sub>

Monitoring Washout  
on B 208.959nm

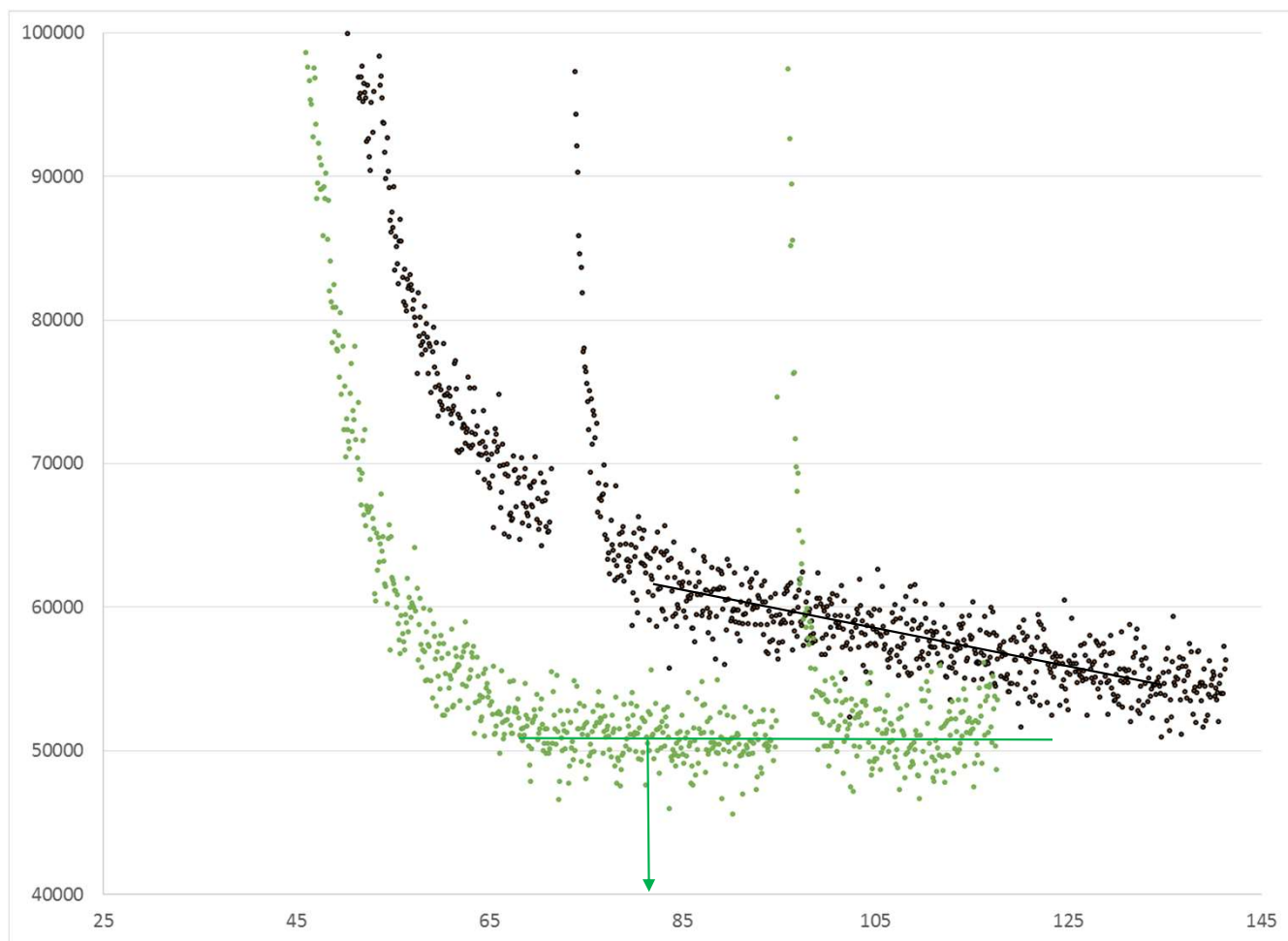
**Rinse w/ 5% HNO<sub>3</sub>**

**Rinse w/ 5% NH<sub>4</sub>OH**

No obvious difference in  
washout at first glance.



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Monitoring Washout  
on B 208.959nm

**Rinse w/ 5%  $\text{HNO}_3$**

**Rinse w/ 5%  $\text{NH}_4\text{OH}$**

**5%  $\text{HNO}_3$**  doesn't go to  
baseline even within 2  
minutes.

**5%  $\text{NH}_4\text{OH}$**  rinses it out  
within 80 seconds.



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# Washout Considerations

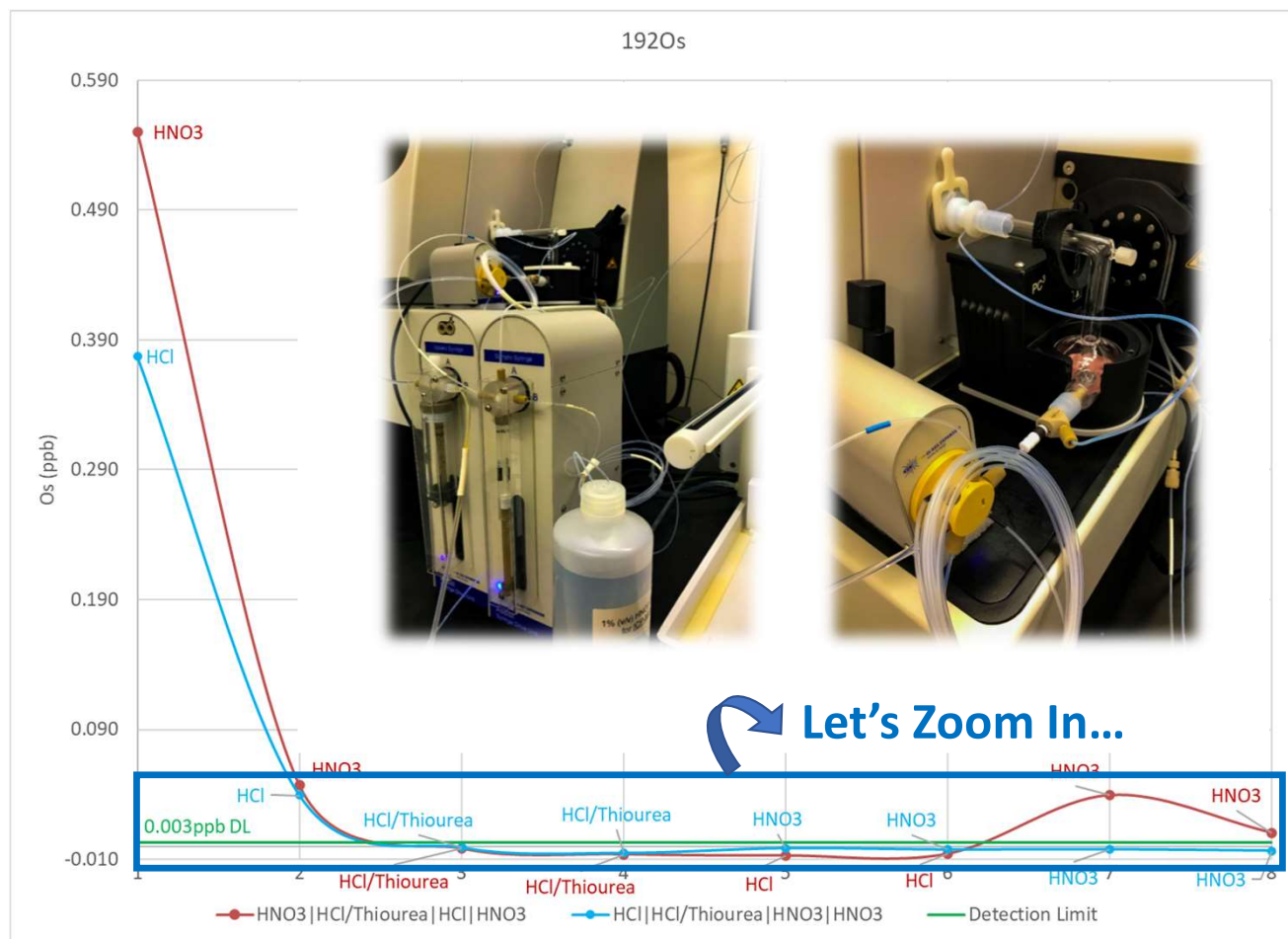
- Boron took much longer to washout when rinsing with the typical 5% v/v  $\text{HNO}_3$  rinse.
- Washout time was significantly reduced using  $\text{NH}_4\text{OH}$ .
- Time graph only tells part of the story.
  - A very small peak is still present when viewing in spectra mode.
  - Time graph can help determine estimated washout.
  - Verify absence of an element by viewing the spectra.



# Experiment with different rinse strategies

- Switching valve and syringe drive systems pose an extra challenge to washout due to the use of a dedicated carrier solution.
- Most ICP-MS users prefer to use  $\text{HNO}_3$  to prevent  $^{40}\text{Ar}^{35}\text{Cl}^+$  &  $^{40}\text{Ar}^{37}\text{Cl}^+$  from interfering on  $^{75}\text{As}$  &  $^{77}\text{Se}$ .
- Running rinses as samples in varying order can help determine which types of rinses are more effective for certain elements and intro-system setups.





50ppb Os run followed by 8 rinses run as samples (Rinses shown)

2 Rinse strategies were performed in pairs

Carrier solution & rinse station contained 1% HNO<sub>3</sub>.

Assist system from GE used with glass double pass spray chamber and Duramist Nebulizer



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## Question 5

•What solution do you use in your auto-sampler rinse station?

- A. Deionized Water
- B. Dilute  $\text{HNO}_3$
- C. Dilute  $\text{HCl}$
- D. Dilute  $\text{HNO}_3/\text{HCl}$
- E. Something Else



# Key Topics

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- **Maintenance recommendations**

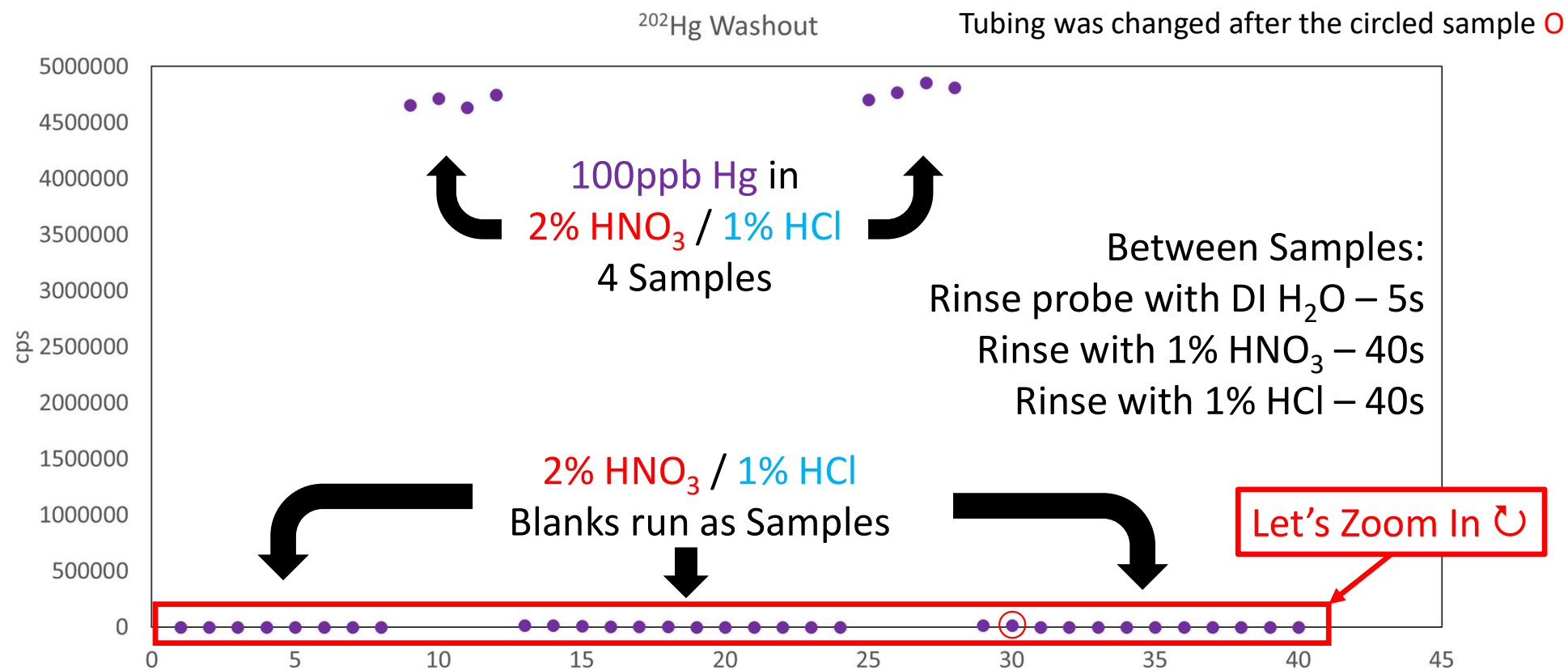


# Peristaltic Pump Tubing

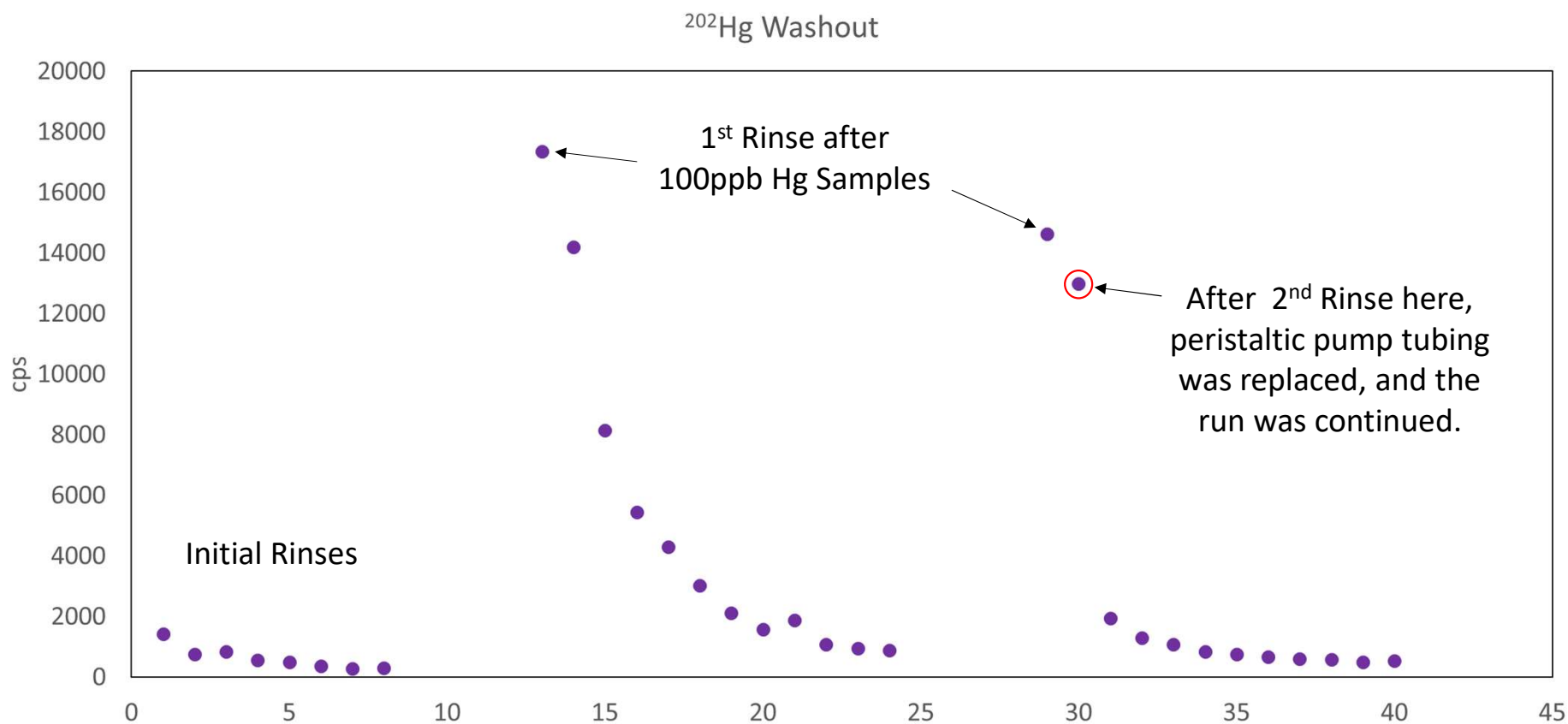
- If using peristaltic pump tubing for sample delivery...
  - Consider changing the tubing daily
  - Or directly after a run with “sticky” elements



# Example of effect of changing tubing



# Example of effect of changing tubing



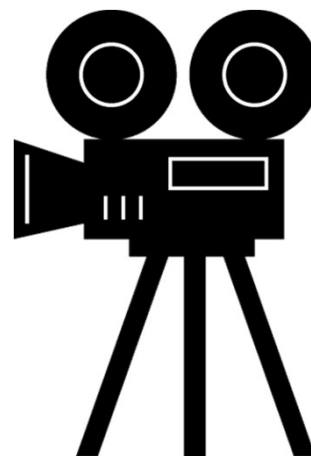
# Switching Valve Systems

- Sample loops do inevitably become dirty
- When not in use, make sure carrier solution flows through the sample loop to continuously clean it
- Replacing valve sleeves can help when routine cleaning no longer helps.



# Switching Valve Systems

- If using a sample loop, shaking the loop while it is rinsing can help clear out stuck elements.
- This phenomenon can be observed using real-time display



## Sample Loop Connections

1<sup>st</sup> time shaking the loop.

2<sup>nd</sup> time shaking the loop.

Less contaminant remains, element likely getting stuck in the connector to the valve.



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# Weekly Soaking of Various Parts

- Soak dirty spray chambers, nebulizers, and valve sleeves in 25% solution of RBS-25.
- Rinse with lots of DI H<sub>2</sub>O.

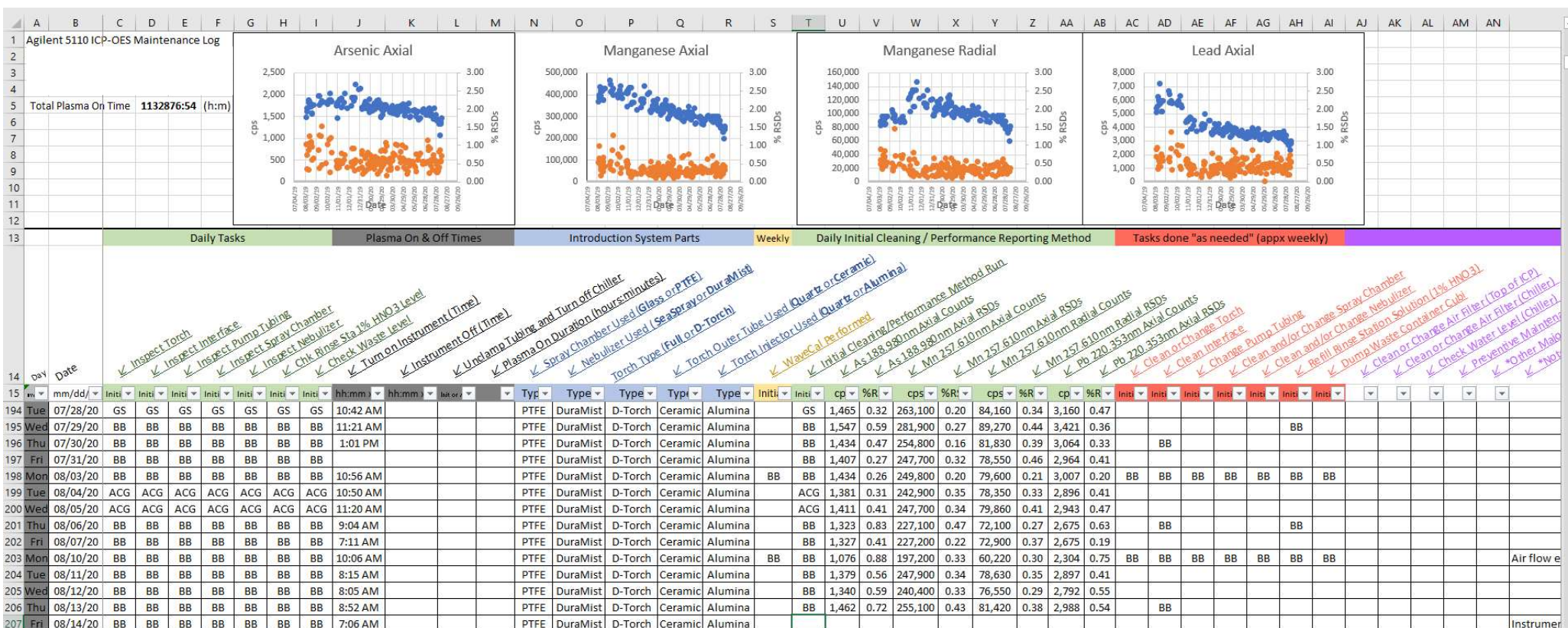


# More tips and tricks for washout issues

- Keep up with routine instrument maintenance to prevent other washout issues.
  - Torch parts, cones/interface, tubing, autosampler probe, rinse station reservoirs
  - Keep a detailed log of maintenance steps and record performance report data

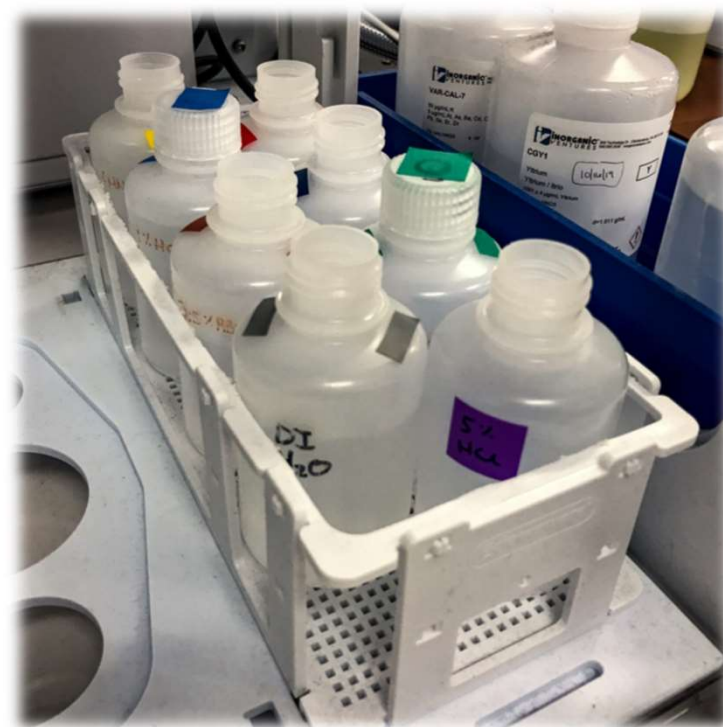


# Example of a Daily Maintenance Log



# More tips and tricks for washout issues

- Keep several different rinse solutions on hand to washout high TDS samples more effectively
  - This will keep your lab running smoothly to get more done!



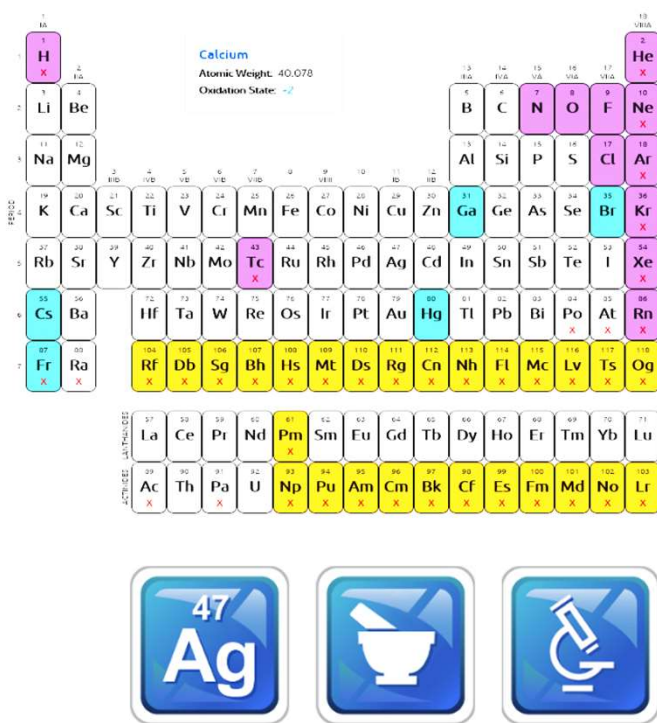
# Final Thoughts

- Identify the elements that give you trouble
- Experiment with different rinse solutions
- Keep up with routine instrument maintenance
- Identify specific causes for washout issues
- Develop methods that include effective rinse strategies and only the elements you require





# Technical Support – Available to Everyone Online Resources at [inorganicventures.com](http://inorganicventures.com)



The image shows a standard periodic table of elements. The element Calcium (Ca) is highlighted in pink. A callout box for Calcium provides the following information:

- Calcium
- Atomic Weight: 40.078
- Oxidation State: +2

Below the periodic table, there are three icons: a blue square with the number 47 and the symbol Ag, a blue square with a mortar and pestle, and a blue square with a microscope.

Customers can visit our website's Tech Center, which includes:

- Interactive Periodic Table
- Sample Preparation Guide
- Trace Analysis Guide
- ICP Operations Guide
- Expert Advice
- And much, much more.



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